Roll No.

Total Pages: 4

1781/M

M-35/2051

ENVIRONMENTAL CHEMISTRY

Paper-401

Semester-IV

Time allowed: 3 Hours [Maximum Marks: 55

Note: The candidates are required to attempt two questions each from section A carrying 8 marks and section B carrying 8½ marks each and the entire Section C consisting of 11 questions carrying 2 marks each.

SECTION-A

- Explain the concept and scope of Environmental Chemistry.
- What do you mean by air pollutants? Discuss in detail the chemistry of air pollutants with suitable examples.

- 3. (a) Write a note on chemical and photochemical reactions taking place in the atmosphere.
 - (b) Describe the development mechanism of acid rain.
- 4. (a) Describe various equipments involved in the monitoring of air pollutants. 4
 - (b) How to ascertain the path of pesticides in environment? Explain? 4

SECTION-B

- 5. (a) Describe various toxic pollutants present in the environment. Also discuss the threshold values of toxic pollutants and the various methods to minimize them. $4\frac{1}{2}$
 - (b) Describe BOD and COD of industrial effluent.

- 6. Discuss principle, theory and instrumentation of Atomic Absorption Spectroscopy. How is it utilized for the analysis of pollution data? 8½
- 7. Write notes on the following:
 - (a) Various columns used in ionchromatography. 4½
 - (b) FTIR and pollution analysis. 4
- 8. (a) Describe the analysis of metal pollutants in effluents. 4
 - (b) Discuss biochemical effects of Hg, Cd, As and Pb. $4\frac{1}{2}$

SECTION-C

- 9. (i) Explain green house effect.
 - (ii) What are environmental segments?
 - (iii) What is meant by photochemical smog?
 - (iv) Describe water standards for drinking water.

- (v) Discuss the role of micro-nutrients in soil.
- (vi) What is meant by coagulation?
- (vii) Explain "Degradation of pesticides."
- (viii) Define Lambert Beer Law.
- (ix) What is the role of detector in Ion-chromatography?
- (x) Discuss the mechanism of scale formation.
- (xi) What do you mean by disinfectants? Explain.

 $2 \times 11 = 22$